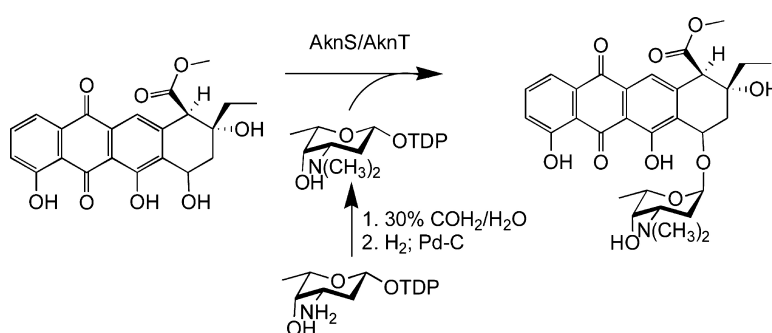


Characterization of Rhodosaminyl Transfer by the AknS/AknT Glycosylation Complex and Its Use in Reconstituting the Biosynthetic Pathway of Aclacinomycin A

Catherine Leimkuhler, Micha Fridman, Tania Lupoli, Suzanne Walker, Christopher T. Walsh, and Daniel Kahne

J. Am. Chem. Soc., 2007, 129 (34), 10546-10550 • DOI: 10.1021/ja072909o • Publication Date (Web): 08 August 2007

Downloaded from <http://pubs.acs.org> on February 15, 2009



More About This Article

Additional resources and features associated with this article are available within the HTML version:

- Supporting Information
- Links to the 2 articles that cite this article, as of the time of this article download
- Access to high resolution figures
- Links to articles and content related to this article
- Copyright permission to reproduce figures and/or text from this article

[View the Full Text HTML](#)

Characterization of Rhodosaminyl Transfer by the AknS/AknT Glycosylation Complex and Its Use in Reconstituting the Biosynthetic Pathway of Aclacinomycin A

Catherine Leimkuhler,[#] Micha Fridman, Tania Lupoli, Suzanne Walker, Christopher T. Walsh, and Daniel Kahne*

Contribution from the Department of Chemistry and Chemical Biology, Harvard University, Cambridge, Massachusetts 02138, Department of Biological Chemistry and Molecular Pharmacology, and Department of Microbiology and Molecular Genetics, Harvard Medical School, 200 Longwood Avenue, Boston, Massachusetts 02115

Received April 25, 2007; E-mail: kahne@chemistry.harvard.edu

Abstract: The tetracyclic core of anthracycline natural products with antitumor activity such as aclacinomycin A are tailored during biosynthesis by regioselective glycosylation. We report the first synthesis of TDP-L-rhodosamine and demonstrate that the glycosyltransferase AknS transfers L-rhodosamine to the aglycone to initiate construction of the side-chain trisaccharide. The partner protein AknT accelerates AknS turnover rate for L-rhodosamine transfer by 200-fold. AknT does not affect the K_m but rather affects the k_{cat} . Using these data, we propose that AknT causes a conformational change in AknS that stabilizes the transition state and ultimately enhances transfer. When the subsequent glycosyltransferase AknK and its substrate TDP-L-fucose are also added to the aglycone, the disaccharide and low levels of a fully reconstituted trisaccharide form of aclacinomycin are observed.

Introduction

AknS is a glycosyltransferase (Gtf) that is involved in the biosynthesis of aclacinomycin A (**1**, Figure 1). Aclacinomycin A is a member of the anthracyclines, a class of microbial secondary metabolites produced by *Streptomyces galilaeus* with antitumor activity. Aclacinomycin A, along with other members of its class, including daunomycin (**2**) and adriamycin (**3**), is used in the clinic to treat various cancers.¹ The mechanism of action of these chemotherapeutics is reported to be induction of apoptosis following binding to double-stranded DNA fragments.² All anthracyclines contain a conserved tetracyclic aglycone, 7,8,9,10-tetrahydro-5,12-naphthacenequinone, with a mono- to trisaccharide moiety attached to the C₇-OH.³ The trisaccharide of aclacinomycin A, which consists of three different deoxy sugars, L-rhodosamine, 2-deoxy-L-fucose, and L-cinerulose A, has been shown to play a key role in binding to the minor groove of target DNA sequences.^{4–6} The trisaccharide is assembled by two Gtfs, AknS, which attaches the first carbohydrate moiety, TDP-L-rhodosamine (**5**, Figure 2), to the (**4**, Figure 2) to yield rhodosaminyl-aklavinone (**6**). AknK then

sequentially attaches TDP-2-deoxy-L-fucose (**7**) and TDP-L-rhodosamine (**8**), producing first 2-deoxy-fucosyl-rhodosaminyl-aklavinone (**9**) and then the trisaccharide, aclacinomycin A (**3**).⁷

The turnover of AknS is substantially increased by the addition of an accessory protein, AknT,⁸ which is encoded by a gene (*aknT*) found directly upstream of the *aknS* gene.⁹ A few other Gtfs involved in the biosynthesis of aminosugar-containing macrolides have also been found to require accessory proteins for good activity: DesVII, which is involved in the biosynthesis of narbomycin and EryCIII, which is involved in the biosynthesis of erythromycin.^{10,11} How these accessory proteins accelerate glycosyltransfer is not known. The AknS/AknT system is ideal for addressing this issue because the aglycon substrate is chromogenic, which facilitates kinetic analysis of the glycosylation reaction. Here we report that AknT accelerates glycosyltransfer by increasing k_{cat} by 2 orders of magnitude; it has no effect on substrate binding.⁸ We propose that AknT facilitates a conformational change in AknS that stabilizes the transition state.

Experimental Section

Materials. AknS, AknT, and AknK were overexpressed and purified as previously described.^{7,8} TDP-daunosamine and TDP-2-deoxy-L-

[#] Current address: Department of Molecular and Cellular Biology, Harvard University, Bauer 307, 7 Divinity Avenue, Cambridge, MA 02138.

- (1) Larsen, A. K.; Escargueil, A. E.; Skladanowski, A. *Pharmacol. Ther.* **2003**, *99*, 167–181.
- (2) Rabbani, A.; Finn, R. M.; Ausio, J. *Bioessays* **2005**, *27*, 50–56.
- (3) Fujii, I.; Ebizuka, Y. *Chem. Rev.* **1997**, *97*, 2511–2524.
- (4) Chaires, J.; Satyanarayana, S.; Suh, D.; Fokt, I.; Przewlaka, T.; Priebe, W. *Biochemistry* **1996**, *35*, 2047–2053.
- (5) Frederick, C.; Williams, L.; Ughetto, G.; van der Marel, G.; van Boom, J.; Rich, A.; Wang, A. *Biochemistry* **1990**, *29*, 2538–2549.
- (6) Temperini, C.; Messori, L.; Orioli, P.; Di Bugno, C.; Animati, F.; Ughetto, G. *Nucleic Acids Res.* **2003**, *31*, 1464–1469.

- (7) Lu, W.; Leimkuhler, C.; Oberthur, M.; Kahne, D.; Walsh, C. T. *Biochemistry* **2004**, *43*, 4548–4558.
- (8) Lu, W.; Leimkuhler, C.; Gatto, G. J., Jr.; Kruger, R. G.; Oberthur, M.; Kahne, D.; Walsh, C. T. *Chem. Biol.* **2005**, *12*, 527–534.
- (9) Raty, K.; Kunnari, T.; Hakala, J.; Mantsala, P.; Ylihonko, K. *Mol. Gen. Genet.* **2000**, *264*, 164–172.
- (10) Yuan, Y.; Chung, H. S.; Leimkuhler, C.; Walsh, C. T.; Kahne, D.; Walker, S. J. *Am. Chem. Soc.* **2005**, *127*, 14128–14129.
- (11) Lee, H. Y.; Chung, H. S.; Hang, C.; Khosla, C.; Walsh, C. T.; Kahne, D.; Walker, S. J. *Am. Chem. Soc.* **2004**, *126*, 9924–9925.

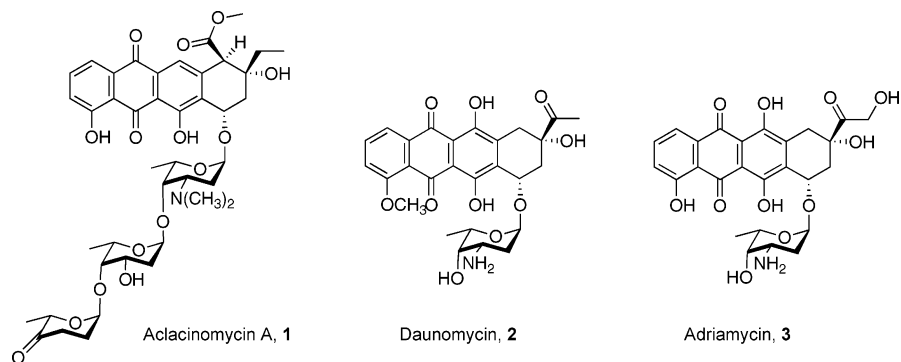


Figure 1. Anthracycline antitumor agents.

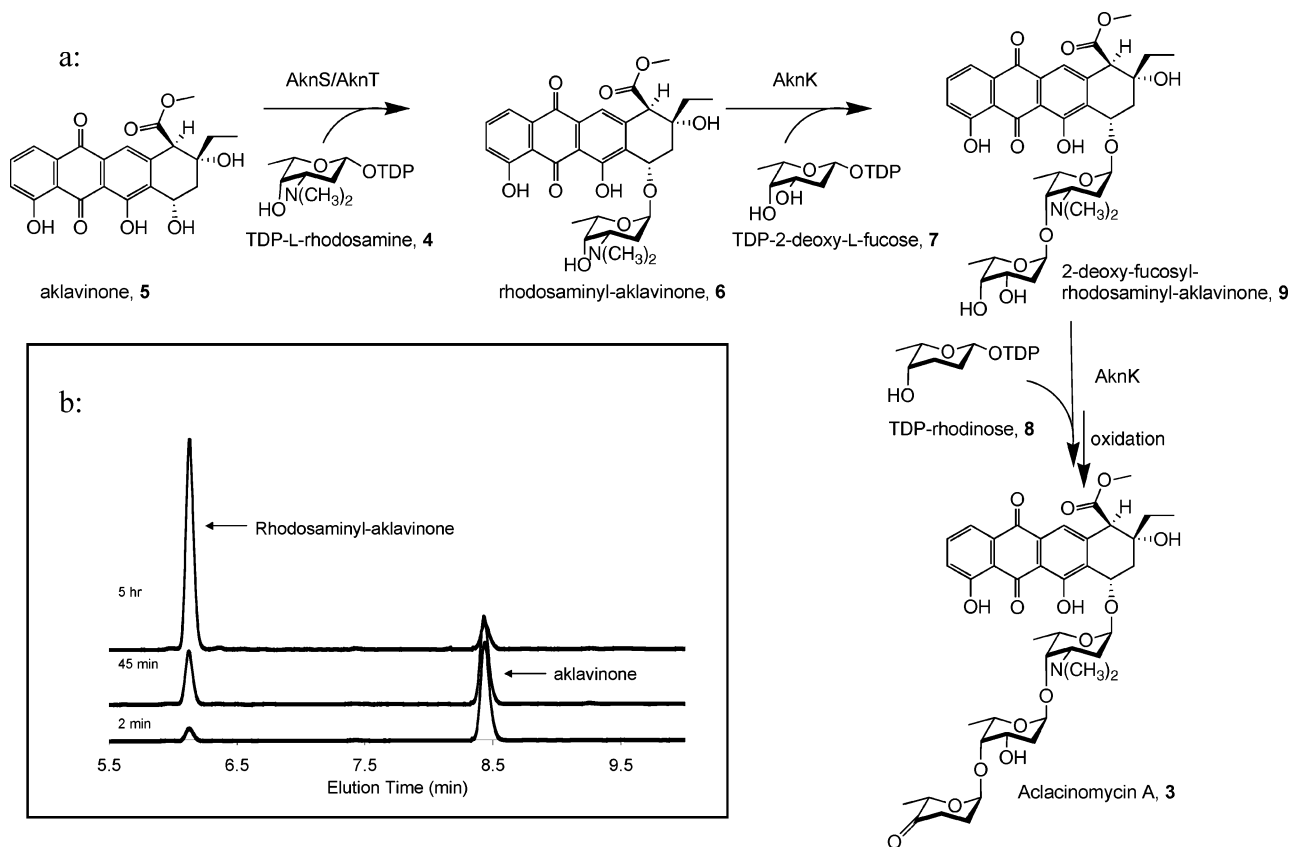


Figure 2. Biosynthetic pathway of aclacinomycin A. (a) Proposed biosynthetic pathway: AknS/AknT transfers TDP-L-rhodosamine to aklavinone **4** to give rhodosaminyl-aklavinone **6**. AknK then transfers TDP-2-deoxy-L-fucose followed by TDP-L-rhodosine which is then oxidized to yield aclacinomycin A. (b) HPLC analysis of AknS/T transfer of TDP-L-rhodosamine to aklavinone **4** to give rhodosaminyl-aklavinone **6**.

fucose were synthesized as previously reported.^{7,12} The aglycone and pseudo-aglycone were obtained through degradation of aclacinomycin A.^{7,8} Other chemicals were purchased from Sigma.

Synthesis of TDP-Rhodosamine. TDP-L-rhodosamine (**5**) was synthesized from known TDP-L-daunosamine (**10**) (Figure 3).^{7,12} A more detailed experimental procedure can be found in the Supporting Information of this paper.

Characterization of the Glycosylation Reaction Catalyzed by AknS/AknT. Aklavinone (**4**) and TDP-L-rhodosamine (**5**) were incubated with AknT and AknS in 50 μ L of reaction buffer (75 mM Tris [pH = 7.5], 10 mM MgCl₂, and 10% [v/v] DMSO) at 25 °C for 1–10 min unless otherwise noted. An aliquot (10 μ L) of the reaction mixture was quenched with 90 μ L of methanol. The sample was analyzed by RP-HPLC using a Phenomenex C18 column (30–100% acetonitrile with 0.1% TFA in water over 8 min, 1 mL/min) (Figure 2). The

products were monitored at 435 nm. The molecular weights of the products were confirmed by ESI-MS. The peak areas for the anthracycline monoglycoside products and the remaining aglycone substrate were integrated, and the product concentration was deduced from its percentage of the total area peak. The initial velocity data were fitted to the Michaelis–Menten equation to obtain K_m and k_{cat} values.

To determine the kinetic parameters for AknS, 0.1 μ M AknS and 0.3 μ M AknT were included in each reaction. For the measurement of K_m for TDP-L-rhodosamine, 0–1600 μ M were used, while the aglycone was kept at 100 μ M. For the measurement of the K_m for the aklavinone, 1.5–200 μ M aglycones were used, while TDP-rhodosamine was maintained at 1600 μ M.

Characterization of the Glycosylation Reaction Catalyzed by AknS. The same general buffer and the procedure for monitoring the reaction were used as described above. In order to determine the kinetic parameters of the reaction, 1 μ M of AknS was used in each reaction.

(12) Oberthur, M.; Leimkuhler, C.; Kahne, D. *Org. Lett.* **2004**, *6*, 2873–2876.

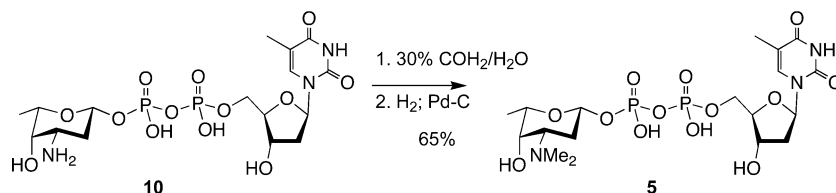


Figure 3. Synthesis of TDP-L-rhodosamine.

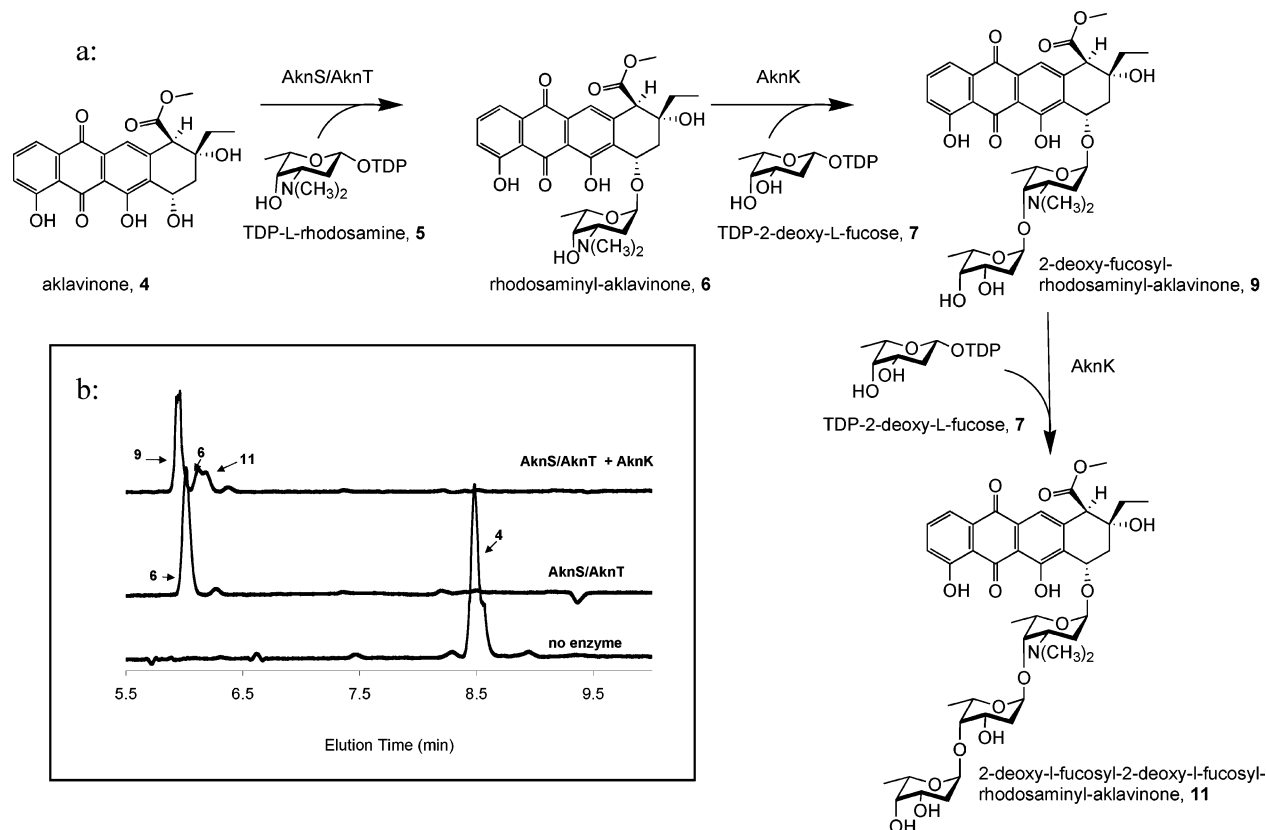


Figure 4. AknS and AknK act in tandem to form a trisaccharide product. (a) Biosynthetic pathway of aclinomycin A. AknS/AknT transfers TDP-L-rhodosamine to aklavinone **4** to give rhodosaminyl-aklavinone **6**. AknK transfers TDP-2-deoxy-L-fucose to yield the disaccharide product **9** and the trisaccharide product transfer **11**. (b) HPLC trace after the addition of each enzyme. The bottom trace shows aklavinone, the middle trace shows the conversion of aklavinone to **6** after addition of AknS/T, and the top trace shows the conversion of **6** to **9** and **11** after addition of AknK.

The AknS/T reaction was also tested at $1 \mu\text{M}$ of AknS/ $3 \mu\text{M}$ of AknT, and we found the kinetic parameters to be comparable to the transfer using $0.1 \mu\text{M}$ of AknS/ $0.3 \mu\text{M}$ AknT. For the measurement of K_m for TDP-L-rhodosamine, 0 – $1600 \mu\text{M}$ were used, while aglycone substrates were kept at $100 \mu\text{M}$. For the measurement of the K_m for the aklavinone, 1.5 – $200 \mu\text{M}$ aglycones were used, while the concentration of TDP-L-rhodosamine was maintained at $1600 \mu\text{M}$.

Reconstitution of Aclinomycin by Tandem Glycosylation with AknS/AknT and AknK. The reconstitution of aclinomycin consisted of two steps. In the first step, $100 \mu\text{M}$ aklavinone (**4**) and $1600 \mu\text{M}$ TDP-L-rhodosamine were incubated with $9 \mu\text{M}$ AknS and $3 \mu\text{M}$ AknT in reaction buffer (75 mM Tris [pH = 7.5], 10 mM MgCl_2 , and 10% [v/v] DMSO: $200 \mu\text{L}$) at 25°C until HPLC analysis showed complete conversion to rhodosamine-aklavinone (**6**) (Figures 2 and 4). A volume of $100 \mu\text{L}$ of this reaction mixture was incubated with $5 \mu\text{M}$ AknK and $1600 \mu\text{M}$ TDP-2-deoxy-L-fucose (**7**) for 6 h to obtain 2-deoxyfucosyl-rhodosaminyl-aklavinone (**8**) and a small amount of 2-deoxyfucosyl-2-deoxyfucosyl-rhodosaminyl-aklavinone (**11**). To the other $100 \mu\text{L}$ of the reaction mixture, only AknK was added; no additional transfer was seen with this reaction (even after overnight incubation), indicating that AknK does not use TDP-rhodosamine as a substrate. The first mixture was analyzed by RP-HPLC using a C18 Phenomenex Column

(30 – 100% acetonitrile in 0.1% TFA in water over 8 min, 1 mL/min). The molecular weights of the desired products were confirmed by ESI-MS (for monoglycosylated calculated 569.6 , found 570.1 [$\text{M} + \text{H}$] $^+$; for diglycosylated calculated 699.7 , found 701.2 [$\text{M} + \text{H}$] $^+$; for triglycosylated calculated 829.9 , found 830.3 [$\text{M} + \text{H}$] $^+$).

Investigation of α -TDP-L-Rhodosamine as Substrate and/or Inhibitor. The AknS/T reaction was performed as described above, but instead of incubating with β -TDP-L-rhodosamine, α -TDP-L-rhodosamine was used. No transfer was seen for the reaction. In order to evaluate the compound as an inhibitor of the reaction, $1600 \mu\text{M}$ β -TDP-L-rhodosamine was incubated with $100 \mu\text{M}$ aklavinone, $0.1 \mu\text{M}$ AknS, and $0.3 \mu\text{M}$ AknT in the reaction buffer described above ($20 \mu\text{L}$ total volume). α -TDP-L-rhodosamine was added to individual reactions in a range of concentrations from 0 to 20 mM . The reactions were monitored as described above.

Results

TDP-L-Rhodosamine Is the Natural Substrate for Akns/AknT. Characterization of the Akns/AknT system was previously carried out using an alternate sugar donor, 2-deoxy-L-fucose presented as the TDP nucleotide.⁸ The turnover for the

Table 1. Kinetic Analysis of AknS/AknT Activity versus AknS Activity

enzymes included in the reaction mixture	K_m (μM)		k_{cat}
	aklavinone	TDP-L-rhodosamine	
AknS and AknT	5.7 ± 0.8	280 ± 20	9.6
AknS	13.0 ± 3	349 ± 70	0.05

fucosylation reaction was very slow. In order to study the role of AknT in facilitating the glycosyltransfer reaction, we required the natural substrate, the identity of which was not known. We suggested in our previous work that AknS might add L-daunosamine from β -TDP-L-daunosamine, the product of which would later be methylated to produce rhodosamine.⁸ However, the catalytic efficiency of an α/β anomeric mixture of TDP-L-daunosamine was found to be poor, suggesting either that α -TDP-L-daunosamine inhibits the reaction or that β -TDP-L-daunosamine is not the natural donor. We began the studies reported here by evaluating the transfer of β -TDP-L-daunosamine purified from the α/β mixture. The enzyme couples the pure β -substrate at the same rate as the α/β mixture (0.05 min^{-1}), making it unlikely that TDP-L-daunosamine is the natural sugar donor substrate of AknS.

We next synthesized β -TDP-L-rhodosamine (**5**, Figure 3) by addition of formaldehyde to β -TDP-L-daunosamine (**10**, Figure 3) followed by hydrogenation. The product is unstable to HPLC purification, and the final purification step was done using size exclusion chromatography under mild basic conditions (1% ammonium bicarbonate in H_2O). The product was characterized by ^1H , ^{31}P NMR, and by HRESI-MS (data reported in the Experimental Section). Following purification, the β -TDP-L-rhodosamine can be stored both in water solution and as the dry ammonium salt at -20°C for a few months.

AknS/AknT transferred rhodosamine from β -TDP-L-rhodosamine to aglycone **4**, as judged by a RP-HPLC assay and ESI-MS of the putative product ($[\text{M} - \text{H}]^+$; calculated 569.3, found 569.3) (Figure 2). In previous work we had established that transfer of TDP-2-deoxy-L-fucose is most efficient using a ratio of 1:3 AknS/AknT.⁸ The optimal AknS/AknT ratio for transfer of TDP-rhodosamine was also 1:3. Under optimal conditions, kinetic parameters measured by monitoring product formation via RP-HPLC were found to be $k_{\text{cat}} = 9.6 \text{ min}^{-1}$; $K_m(\text{aglycone}) = 5 \mu\text{M}$; $K_m(\text{TDP-rhodosamine}) = 280 \mu\text{M}$ (Table 1). The k_{cat} value is 2 orders of magnitude faster than that for TDP-2-deoxy-L-fucose and 3 orders of magnitude faster than that for TDP-L-daunosamine, supporting the proposal that TDP-L-rhodosamine is the natural substrate of AknS.

We next tested the ability of AknS to transfer TDP-L-rhodosamine to the aglycone **4** in the absence of AknT. The k_{cat} for glycosyl transfer is 0.05 min^{-1} , and the K_m values for the aglycone and the TDP-rhodosamine substrates were 12.8 and $349 \mu\text{M}$, respectively (Table 1). Thus, while there is no substantial change in substrate binding in the absence of AknT, there is a dramatic change in k_{cat} . The presence of AknT accelerates turnover mediated by AknS by 2 orders of magnitude.

AknK Does Not Transfer TDP-L-Rhodosamine. AknK is the next Gtf in the biosynthesis of aclacinomycin A, and its natural substrate is TDP-2-deoxy-L-fucose. To probe the substrate selectivity of AknK, we incubated it with the pseudo-aglycone **6** and TDP-L-rhodosamine. No transfer was observed

after 24 h of incubation. Therefore, although AknS accepts TDP-2-deoxy-L-fucose as a slow substrate, AknK does not utilize TDP-L-rhodosamine.

Reconstitution of the Biological Pathway. AknS/T and AknK were used sequentially to synthesize 2-deoxy-fucosyl-rhodosamine-aklavinone from the aglycone and the corresponding TDP sugars. After incubating for 24 h, two products were observed by LC-MS: the disaccharide aklavinone, 2-deoxy-fucosyl-rhodosaminyl-aklavinone **9**, and the trisaccharide aklavinone, 2-deoxy-fucosyl-2-deoxy-fucosyl-rhodosaminyl-aklavinone **11** (Figure 4). Thus, the tandem transfer successfully produced the disaccharide product as the major product, but minor amounts of the trisaccharide product were observed, consistent with previous observations that AknK can add 2-deoxy-L-fucose in two sequential reactions to yield a trisaccharide product.⁷

Discussion

Prior to this study AknS and AknT had been overexpressed and purified,⁸ but the glycosyltransfer reaction was not characterized with the natural substrate. Instead, β -TDP-2-deoxy-L-fucose and α/β -TDP-L-daunosamine were used as substrates because they were readily available. The AknS/AknT transferase activity was extremely low when TDP-2-deoxy-L-fucose was used (0.22 min^{-1}) and even lower when α/β -TDP-L-daunosamine was used (0.01 min^{-1}). Therefore, although we were able to show that AknT enhances the activity of AknS, the donor substrates were too poor to permit accurate kinetic measurements in the absence of AknT, which made it difficult to address the basis for the observed enhancement. In this paper, we have shown that β -TDP-L-rhodosamine is a highly active sugar donor for AknS, and the rates of the glycosyltransfer reaction in the presence and absence of AknT are sufficient to assess the kinetic basis for the observed enhancement.

Here we have reported that AknS/AknT transfers TDP-rhodosamine to the aglycone **4** with a k_{cat} of 9.6 min^{-1} , which is in the range of turnover numbers with other characterized Gtfs of this class (e.g., AknK and NovM).^{7,13} The K_m values were found to be 280 and $5.7 \mu\text{M}$ for TDP-L-rhodosamine and aklavinone **4**, respectively. In the absence of AknT, the K_m for TDP-L-rhodosamine is $329 \mu\text{M}$ and the aklavinone K_m is $12 \mu\text{M}$, whereas the k_{cat} is 0.05 min^{-1} , which is 2 orders of magnitude slower than when AknT is present. These results reveal a negligible influence of AknT on AknS affinity for substrates, as judged by the similar K_m 's, but a dramatic influence of AknT on the catalytic rate constant for the reaction.

AknS is not the only Gtf that uses an accessory protein, as studies of two other homologous systems have been reported. For the EryCIII/EryCII system, it was observed that transient incubation of EryCII with EryCIII converts EryCIII from an inactive form to an enzyme that remains capable of catalyzing glycosyltransfer even after EryCII is removed.^{10,11} Preliminary experiments with the DesVII/DesVIII system have also shown that DesVIII is only required for initial activation of DesVII and that the activated DesVII can catalyze the glycosyl transfer alone.¹⁴ In contrast, significant AknS activity requires a 3-fold excess of AknT during the glycosyl transfer reaction. It has been

(13) Freil Meyers, C. L.; Oberthur, M.; Anderson, J. W.; Kahne, D.; Walsh, C. T. *Biochemistry* **2003**, *42*, 4179–4189.

suggested that Gtf accessory proteins may function as substrate carriers and as chaperones that facilitate the proper folding of their cognate Gtfs. The results reported here are not consistent with a substrate-carrier function for AknT and show instead that this protein markedly affects the turnover number. It is worth mentioning that there is one well-characterized example of a eukaryotic Gtf that functions with another protein. This Gtf, β 1,4-galactosyltransferase-1 (β 4-Gal-T1), interacts with α -lactalbumin (LA).¹⁵ The binding interaction stabilizes a particular conformation of Gal-T1 such that it accelerates the rate of transfer of sugars that are otherwise poor substrates.¹⁶ Detailed kinetic analysis of this system have revealed substantial changes (1000-fold) in the K_m of Gal-T1 for glucose in the presence or absence of LA.¹⁷ Since we do not observe a change in the K_m of AknS for either substrate, aklavinone or TDP-L-rhodamine, in the presence of AknT, AknT may interact with AknS in such a way as to allow AknS to access a more favorable conformation in the transition state. AknS and AknT do not copurify, nor do they form a detectably stable complex, so the acceleratory action must occur in a transient ternary complex of AknS, AknT, and the two substrates.

- (14) Borisova, S. A.; Zhang, C.; Takahashi, H.; Zhang, H.; Wong, A. W.; Thorson, J. S.; Liu, H. W. *Angew. Chem., Int. Ed.* **2006**, *45*, 2748–2753.
(15) Brew, K.; Vanaman, T. C.; Hill, R. L. *Proc. Natl. Acad. Sci. U.S.A.* **1968**, *59*, 491–497.
(16) Ramakrishnan, B.; Qasba, P. K. *J. Mol. Biol.* **2001**, *310*, 205–218.
(17) Klee, W. A.; Klee, C. B. *Biochem. Biophys. Res. Commun.* **1970**, *39*, 833–841.

Conclusion

We have reported the first synthesis of TDP-L-rhodamine and have presented evidence that TDP-L-rhodamine is the natural substrate for AknS during the biosynthesis of aklavinone. Tandem sugar transfer from TDP-L-rhodamine followed by TDP-2-deoxy-L-fucose was accomplished using AknS/T and AknK, respectively. A small amount of trisaccharide product was observed, but the disaccharide product was the major product, consistent with the proposal that TDP-L-rhodamine rather than TDP-2-deoxy-L-fucose is the natural substrate for the third glycosyltransferase.⁹

Access to the natural L-sugar nucleotide substrate allowed us to probe the basis for AknS/T activation, and our results have shown that AknT influences the catalytic rate constant for glycosyltransferase but has no effect on K_m values. The basis for this is under investigation.

Acknowledgment. This work was supported by the NIH (Grant GM076710 to D.K. and S.W. and Grant GM49338 to C.T.W.). C.L. was supported by a Bristol-Myers Squibb Graduate Fellowship in Synthetic Organic Chemistry.

Supporting Information Available: Experimental procedures and spectral data for all compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

JA072909O